

10–100 mg/L NH<sub>3</sub>-N High Range plus

TNTplus<sup>®</sup> 837—Method 10205

**Scope and application:** For municipal and industrial wastewaters, environmental waters and watershed protection monitoring.



## Test preparation

### Test storage

Storage temperature: 2–8 °C (35–46 °F)

### pH/Temperature

The pH of the water sample must be between pH 4–9.

The temperature of the water sample and reagents must be 20 °C (68 °F).

### Before starting

**In case of not working at the correct recommended temperature an incorrect result may be obtained.**

Analyze the samples as soon as possible for best results.

Time dependency:

The final absorbance is reached after a reaction time of **15 minutes** and then remains **constant for a further 15 minutes**.

Review safety information and expiration date on the package.

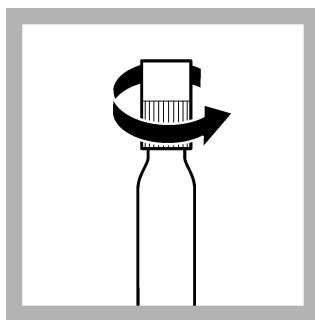
Review the Safety Data Sheets (MSDS/SDS) for the chemicals that are used. Use the recommended personal protective equipment.

Dispose of reacted solutions according to local, state and federal regulations. Refer to the Safety Data Sheets for disposal information for unused reagents. Refer to the environmental, health and safety staff for your facility and/or local regulatory agencies for further disposal information.

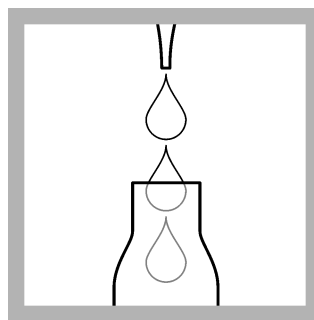
### Procedure



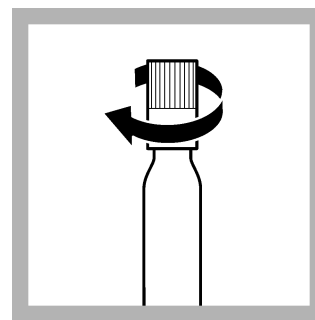
1. Carefully remove the foil from the screwed-on DosiCap Zip.



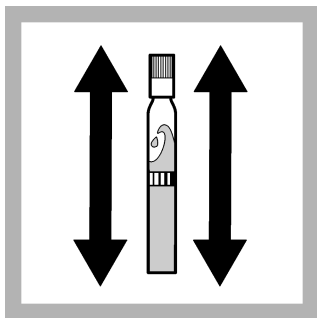
2. Unscrew the DosiCap Zip.



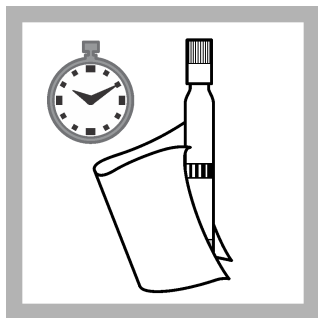
3. Carefully pipet 0.1 mL of sample.



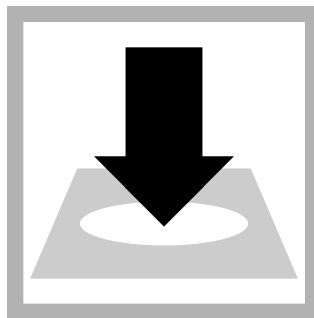
4. **Immediately** screw the DosiCap Zip back on; **fluting at the top**.



5. Shake **vigorously** (2–3 times).



6. After **15 minutes**, thoroughly clean the outside of the vial and evaluate.



7. Insert the vial into the cell holder.  
DR1900: Go to **LCK/TNTplus methods**.  
Select the test, push **READ**.

## Interferences

The ions listed in the table have been individually checked against the given concentrations and do not cause interference. The cumulative effects and the influence of other ions have not been determined.

Primary amines are also determined and cause high-bias results. A 10,000-fold excess of urea does not interfere. All reducing agents interfere and cause low-bias results.

**A large excess of ammonium can cause result displays within the measuring range. It is advisable to carry out a plausibility check by making dilutions.**

The measurement results must be subjected to plausibility checks (dilute and/or spike the sample).

Interference level	Interfering substance
1000 mg/L	Cl <sup>-</sup> , SO <sub>4</sub> <sup>2-</sup>
500 mg/L	K <sup>+</sup> , Na <sup>+</sup> , Ca <sup>2+</sup>
50 mg/L	CO <sub>3</sub> <sup>2-</sup> , NO <sub>3</sub> <sup>-</sup> , Fe <sup>3+</sup> , Cr <sup>3+</sup> , Cr <sup>6+</sup> , Zn <sup>2+</sup> , Cu <sup>2+</sup> , Co <sup>2+</sup> , Ni <sup>2+</sup> , Hg <sup>2+</sup>
25 mg/L	Fe <sup>2+</sup>
10 mg/L	Sn <sup>2+</sup>
5 mg/L	Pb <sup>2+</sup>
2 mg/L	Ag <sup>+</sup>

## Summary of method

Ammonium ions react at pH 12.6 with hypochlorite ions and salicylate ions in the presence of sodium nitroprusside as a catalyst to form indophenol blue.

**TNT<sup>plus</sup>**



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